



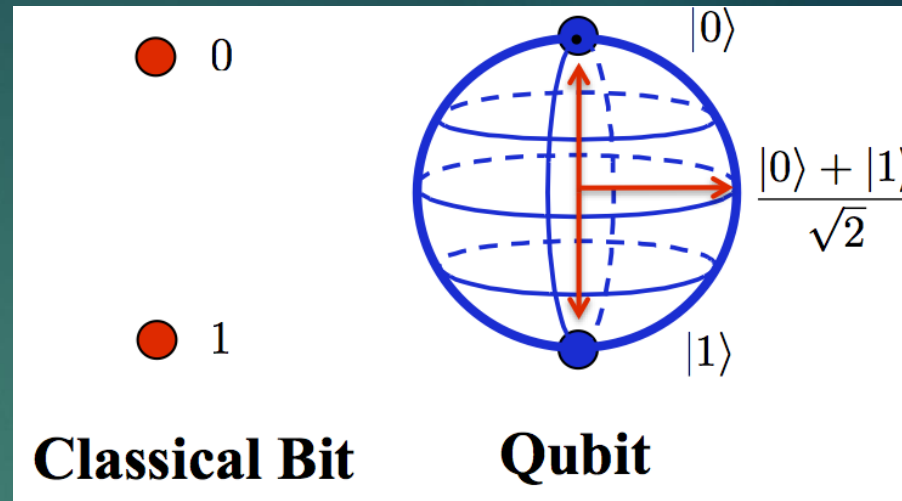
# Toward optically addressing single rare-earth ions in solid state

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# Motivation

- ▶ Rare earth ion as a candidate For quantum bits (qubit)
- ▶ Strong dipole-dipole interaction
- ▶ Long coherence time



The Periodic Table of the Elements

138.9054	57	140.116	58	140.9076	59	144.242	60	(145)	61	150.36	62	151.964	63	157.25	64	158.9253	65	162.500	66	164.9303	67	167.259	68	168.9342	69	173.054	70
58.1	1.10	59.4	1.12	52.7	1.13	58.3	1.14	54.0		54.5	1.17	54.7		59.4	1.20	56.8		57.0	1.22	58.1	1.23	58.3	1.24	59.7	1.25	80.3	
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb														
Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium														
[Xe] 5d <sup>1</sup> 6s <sup>2</sup>	[Xe] 4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup>	[Xe] 4f <sup>3</sup> 6s <sup>2</sup>	[Xe] 4f <sup>4</sup> 6s <sup>2</sup>	[Xe] 4f <sup>5</sup> 6s <sup>2</sup>	[Xe] 4f <sup>6</sup> 6s <sup>2</sup>	[Xe] 4f <sup>7</sup> 6s <sup>2</sup>	[Xe] 4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup>	[Xe] 4f <sup>7</sup> 6s <sup>2</sup>	[Xe] 4f <sup>9</sup> 6s <sup>2</sup>	[Xe] 4f <sup>11</sup> 6s <sup>2</sup>	[Xe] 4f <sup>12</sup> 6s <sup>2</sup>	[Xe] 4f <sup>13</sup> 6s <sup>2</sup>	[Xe] 4f <sup>14</sup> 6s <sup>2</sup>	[Xe] 4f <sup>14</sup> 6s <sup>2</sup>													

# The Rare-earth (RE)- Europium

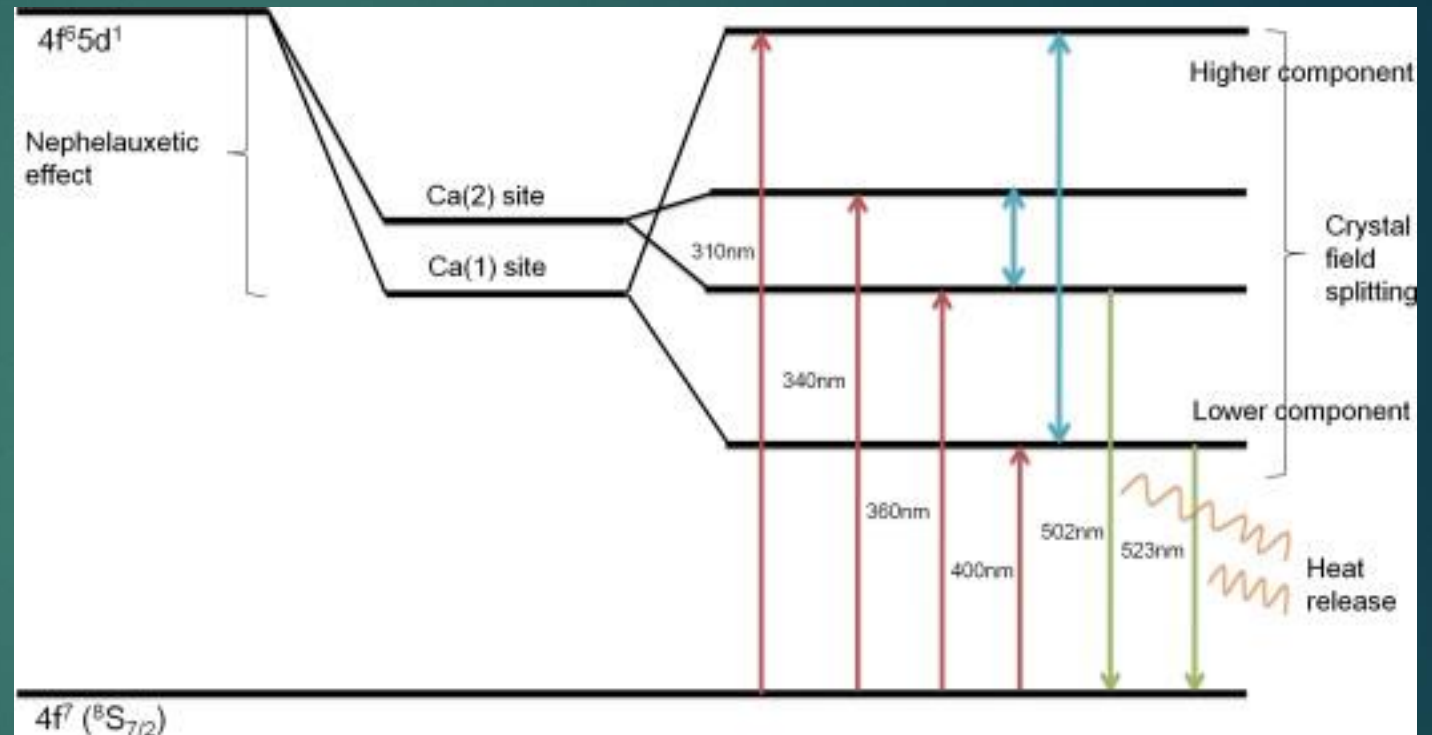
- ▶  $\text{Eu}^{2+}$
- ▶ In  $\text{Ca}_2\text{SiO}_4$

$^{40}\text{Ca}$	$^{44}\text{Ca}$	$^{42}\text{Ca}$	$^{48}\text{Ca}$
$^{28}\text{Si}$	$^{29}\text{Si}$	$^{20}\text{Si}$	$^{28}\text{Si}$
$^{16}\text{O}$	$^{18}\text{O}$	$^{17}\text{O}$	

$\text{Eu}^{2+}$ :  $\text{Ca}_2\text{SiO}_4$  is well studied at 0.1~1 mol % concentration

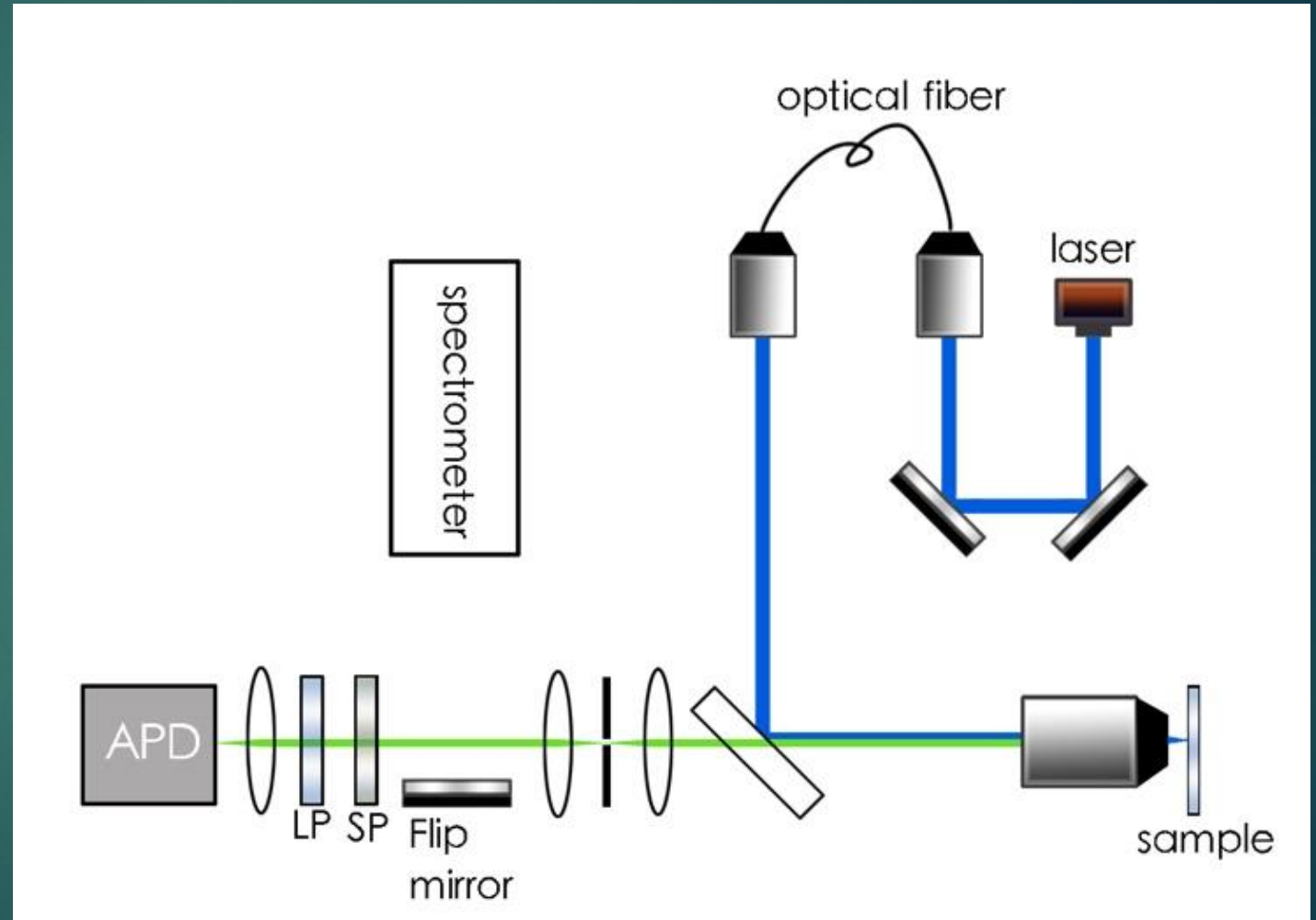
# The energy level

- ▶  $4f^65d \ ^8S_{7/2} \rightarrow 4f^7$  transition
- ▶ Crystal splitting
- ▶ Occupying either Ca(1) or Ca(2) Site



# Setup

- ▶ Diode laser (405nm)
- ▶ Short pass 625nm
- ▶ Long pass 450 nm



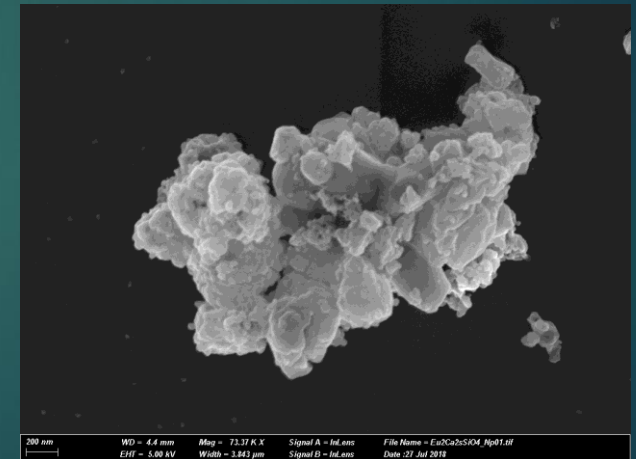
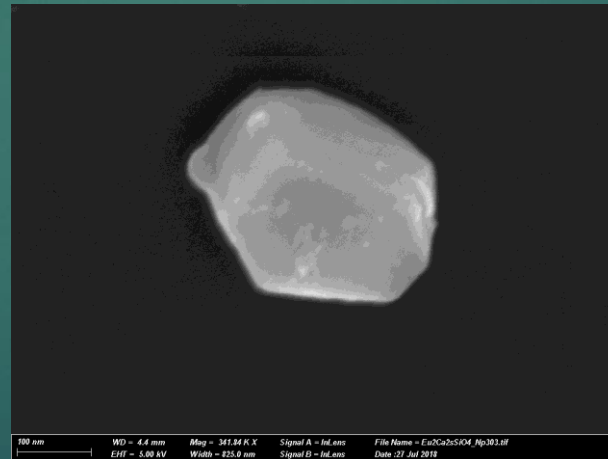
# Crystal producing and Ion stabilization

- ▶ Solid State Reaction
- ▶ Trivalent to divalent
- ▶ Reduced atmosphere



# Crystal producing and Ion stabilization

- ▶ Bad crystal that bleached or blinked
- ▶ Under Scanning electron microscope
- ▶ Annealing temperature matters

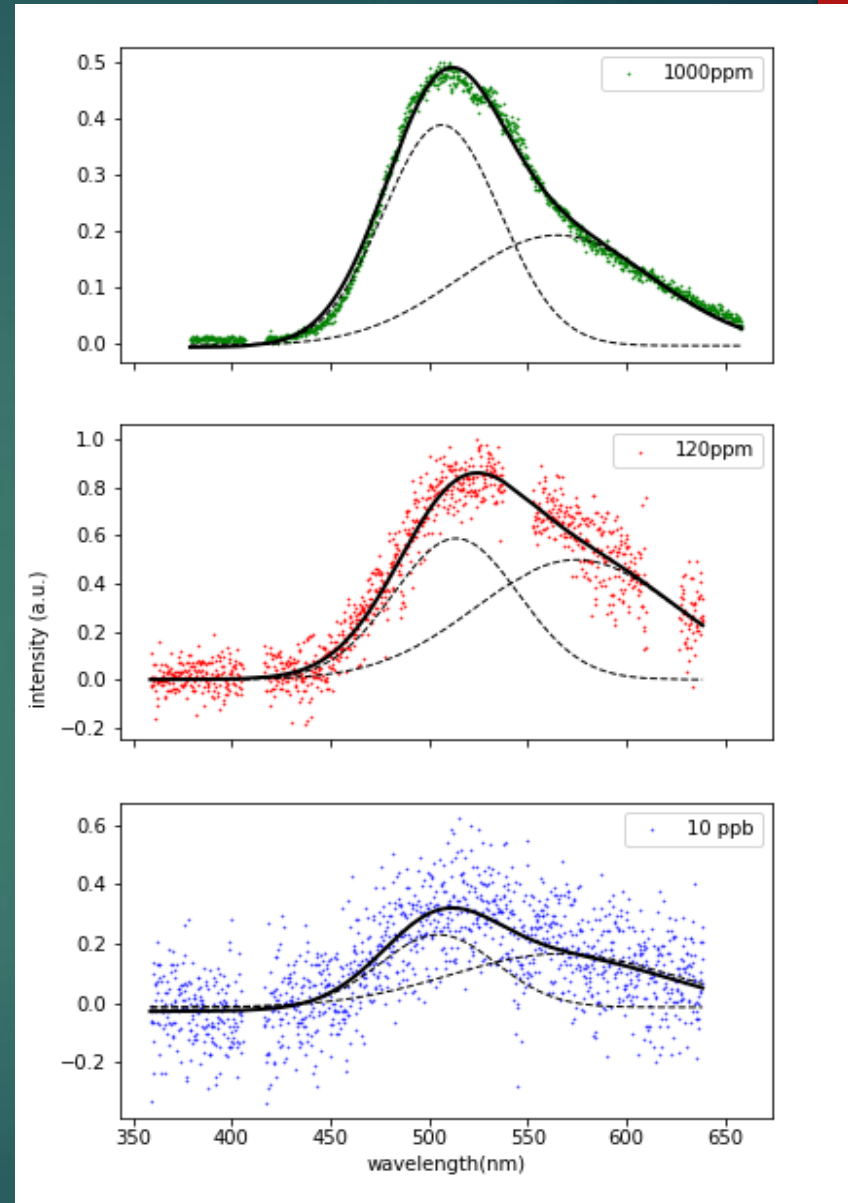


# Spectrum

- ▶ Broad (multi)band
- ▶ Two Gaussian profile

## Red-Shift!

	Ca(1)	Ca(2)
1000 ppm	562.28(0)	504.76(7)
120 ppb	575.75(6)	510.96(0)
10 ppb	595.92(4)	514.76(0)

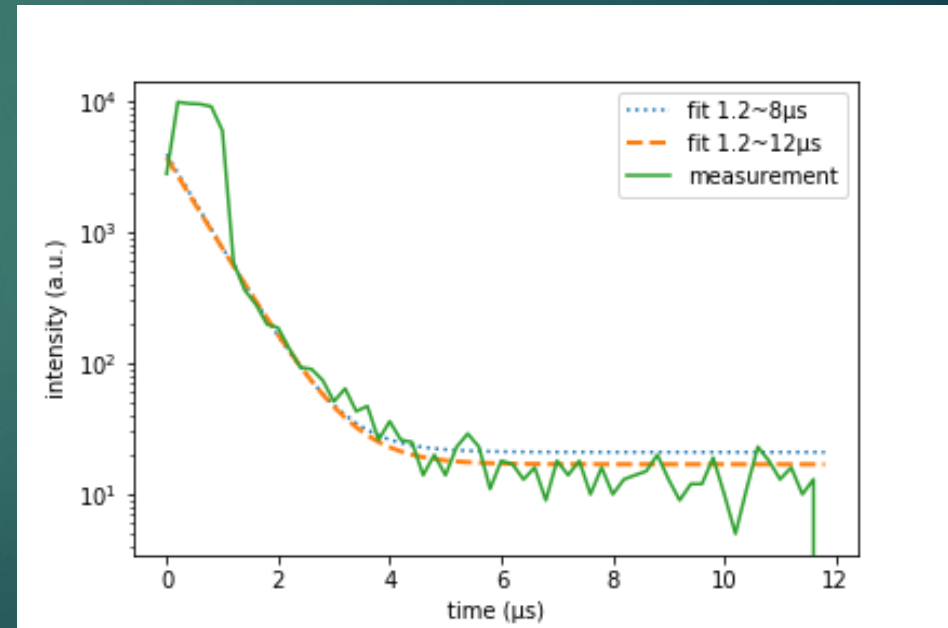




# Fluorescence Life time

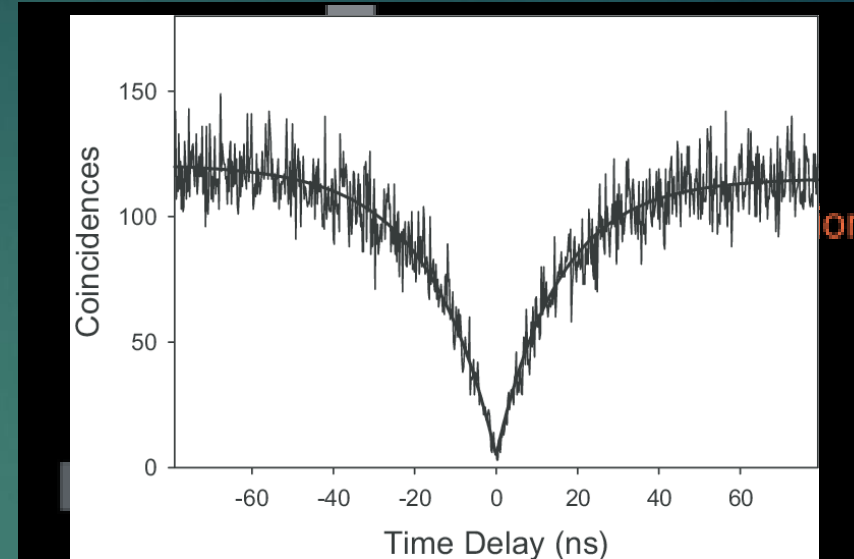
- ▶  $0.62(1) \mu\text{s}$
- ▶ All photons within the wavelength window is collected
- ▶ longer than the previously reported fluorescence lifetime ( $0.53 \mu\text{s}$ )
- ▶ As expected...

**But an unknown fast-decaying source  
Is causing trouble!**



# Future work

- ▶ Anti-bunching measurement
- ▶ Increase the yield of good crystals
- ▶ OR do ion implantation instead



# Future work

- ▶ As for verifying that the nano crystal indeed contain something close to SINGLE Eu<sup>2+</sup> befor setting up the Hanbury Brown and Twiss...
- ▶ Maybe we need photon statistics...
- ▶ How about just look at the spectrum?

$$P(n) = 2^{(-n+1)}$$

# Future work

- ▶ Even if we do not reach single ion level... (proposed by Dr. Kolesov)
- ▶ Find the optical transition at cryogenic temperature

# Many thanks to...

- ▶ Prof. Mingchung Chu, Prof. Sen Yang and Prof. Dajun
- ▶ Prof. Dr. Jörg Wrachtrup, Dr. Roman
- ▶ Thomas Kornher, Stephan Hirschmann, Dr. Rolf Reuter, and Ferdinand Schiller
- ▶ \$\$\$ the Department of Physics, CUHK \$\$\$

Thank you for listening!

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Q&A time